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PRELIMINARY NOTE

Functionalized Perfluorophenyl Azides: New Reagents for Photoaffinity Labeling

JOHN F. W. KEANA* and SUI XIONG CAI

Department of Chemistry, University of Oregon, Eugene, OR 97403 (U.S.A.)

SUMMARY

Several substituted perfluorophenyl azides capable of attachment to other molecules by an acylation reaction were synthesized for use as photoaffinity labeling reagents.

Photoaffinity labeling [1] is widely used to study the proximity of components within biological systems. Photolysis of the photolabel generates a nitrene or carbene intermediate which then reacts rapidly with nearby molecules to form a covalent bond. C-H bond insertion is a desirable reaction pathway. Therefore, it is important to have available photolabels which lead to efficient C-H insertion. Aryl azides are widely used as photolabels. Remarkably, the parent compound phenyl azide gives almost no C-H insertion products in hydrocarbon solvents [2] but instead mostly undergoes ring expansion to dehydroazepine [3].

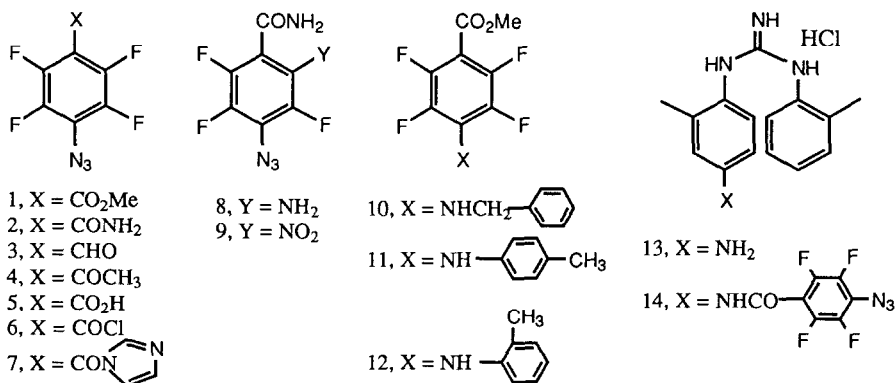
Perfluoroaryl azides may constitute an improved series of photolabels [4] on the basis that a) thermal decomposition of 4-azido-2,3,5,6-tetrafluoropyridine in cyclohexane gave the C-H insertion product in 45% yield [5], and b) photolysis [6] of pentafluorophenyl azide in toluene gave C-H insertion products in 52% yield. Herein, we describe the synthesis of several perfluorophenyl azides that also contain a reactive group for convenient attachment of the new photolabels to other molecules.

Pentafluorophenyl compounds containing an electron withdrawing group such as $-\text{CF}_3$ [7] or $-\text{CN}$ [8] are known to undergo nucleophilic aromatic substitution nearly exclusively at the 4-position [9]. Azide ion was the nucleophile in a few cases [10].

A solution of methyl pentafluorobenzoate (1.19 g, 5.26 mM) and sodium azide (344 mg, 5.29 mM) in 3:1 acetone-water (13 mL) was refluxed for 7 h. The solution

was then cooled, diluted with water (5 mL), and extracted with ether. The extract was dried (Na_2SO_4) and evaporated, giving 1.14 g (87%) of azide **1**, mp 54-55 °C. Similarly prepared in >80% yield were azido benzamide **2**, mp 164-165 °C, azido benzaldehyde **3**, mp 44-45 °C, and azido acetophenone **4**, bp 50 °C/ 0.1 mm. Selective hydrolysis ($\text{NaOH}/\text{MeOH}/\text{H}_2\text{O}$, 25 °C, 8 h, 95%) of ester **1** gave azido acid **5**, mp 140-141 °C. Acid **5** was converted into acid chloride **6** (refluxing $\text{SOCl}_2/\text{CH}_2\text{Cl}_2$, 48 h, 70%; bp 47-49 °C/0.01 mm) and acylimidazole **7** (1,1-carbonyldiimidazole/THF, 1 h, 25 °C, 60%; mp 63-64 °C). Treatment of **2** with NH_3 -saturated dry DMSO at 75 °C for 5 days gave amine **8** (50%), mp 160-161 °C. Oxidation of **8** by $\text{CF}_3\text{CO}_3\text{H}$ [11] gave nitro azide **9**, mp 135-136 °C.

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- For compounds **1-7**, the ^{19}F NMR spectrum showed the expected AA'XX' pattern.
- For new compounds **1-9** and **14**, satisfactory IR, 300 MHz ^1H and 338.7 MHz ^{19}F NMR spectra together with either a C,H,N microanalysis or else a high resolution mass spectrum were obtained. UV max nm (log ϵ) (EtOH): **1**, 263 (4.17); **2**, 255 (4.30); **3**, 250 (4.11); **4**, 267 (4.23); **5**, 259 (4.07); **8**, 265 (3.89), 334 (3.23); **9**, 250 (4.53), sh 300 (3.84).
- For amine **8**, the position of the amino group was indicated by the presence of intramolecular H-bonding between the carbonyl group and the amino group in the IR spectrum. Also, the ^{19}F NMR spectrum showed that there was only one F atom (F-6) coupled to the carboxamide protons.

Photolysis (350 nm) of ester **1** (4.2×10^{-3} M) in toluene at 25 °C gave a mixture of formal C-H insertion products (**10**, 16%; **11**, 9%; **12**, 13%) in 38% yield. The products were separated by preparative TLC, identified by ^1H and ^{19}F NMR and mass spectrometry, and the structures were confirmed by independent synthesis. This result confirms that certain functionalized perfluoroaryl azides also show significant C-H insertion upon photolysis. Attachment of one of these new photolabels to another molecule is illustrated by the selective acylation of amino derivative **13** [12] of the mammalian brain sigma receptor ligand N,N'-di-*o*-tolylguanidine (DTG) [13] with acid chloride **6** to give amide **14**. This ligand shows an $\text{IC}_{50} = 25$ nanomolar against [^3H]-DTG [14]. This value is comparable to that of DTG itself and suggests that **14** may be a useful companion to photolabels already developed for the sigma receptor [12]. Other applications of these new photolabels is underway in our laboratory.

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- 14 This experiment was carried out by Dr. Eckard Weber, Oregon Health Sciences University, Portland, OR.